QF	P CODE: D2BPH2402	(Pages: 2)	Reg. N	o :	
			Name	• •••••	
	SECOND SE	MESTER FYUGP EXAMINA	TION, APRIL	2025	
		MINOR COURSE			
	PHY2MN102	: MODERN PHYSICS AND N	IUCLEAR PH	IYSICS	
		(Credits: 4)			
Tir	me: 2 Hours			Maxin	num Marks: 70
	Answer the followin	Section A g questions. Each carries 3	marke (Coili	na: 21 m	arke)
1.		? Write an expression for the v	•	BL1	CO2
2.	Calculate the value of Com	alculate the value of Compton wavelength. Also find the unit.		BL2	CO1
3.	A source operating at a frequency of 100 MHz radiates a power of 100 kW. Calculate the number of quanta of energy emitted per second.		BL3	CO1	
4.	What is atomic spectra? Differentiate between emission line spectra and absorption line spectrum.		ine spectra	BL2	CO3
5.	0 , , , ,	particles pass through gases ans. To what conclusion about a ion lead?		BL2	CO3
6.	What are the merits of liqu	d drop model?		BL2	CO4
7.	Why ₉₂ ²³⁸ U is not suitable for nuclear fission?			BL2	CO6
8.	What do you mean by a cor	npound nucleus? How are the	ey formed?	BL2	CO6
9.	Give the fundamental laws	of radioactivity.		BL1	CO5
10.	What is the mean life of a ra	adioactive isotope?		BL1	CO5
		Section B			
	Answer the followin	g questions. Each carries 6	marks (Ceili	ng: 36 N	larks)
11.	Calculate the minimum ene	rgy of radiation for pair produc	ction. (PTO)	BL3	CO2

12.	Derive an equation to find the total energy of hydrogen atom.	BL2	CO3		
13.	The longest wavelength in the Lyman series is 121.5 nm and the shortest wavelength in the Balmer series is 364.6 nm. Use the figures to find the longest wavelength of light that could ionize hydrogen.	BL3	CO3		
14.	Explain Bohr model of atom.	BL2	CO3		
15.	Deuteron is the nucleus of heavy hydrogen or deuterium $({}_{1}{}^{2}H)$. Its mass is 3.344 x 10 ⁻²⁷ Kg. It has one proton and one neutron. The mass of proton is 1.673 x 10 ⁻²⁷ Kg and that of neutron is 1.673 x 10 ⁻²⁷ Kg. From these data, find out the mass defect, and binding energy per nucleon.		CO4		
16.	Discuss the nuclear structure.	BL1			
17.	Establish that magic numbers evolve naturally from the shell model.	BL2	CO4		
18.	A nucleus emits an α particle followed by two β particles. Show that the final nucleus is an isotope of the original one.	BL3	CO5		
	Section C				
	Answer any one question. Each carries 10 marks (1 x 10 =	= 10 Marks)			
19.	State the laws of photoelectric effect. Explain how it is accounted on the basis of quantum theory of radiation and hence obtain Einstein's photoelectric equation.	BL2	CO1		
20.	What are the differences and similarities between nuclear fission and nuclear fusion?	BL2	CO6		
	CO : Course Outcome				
	BL : Bloom's Taxonomy Levels (1 – Remember, 2 – Understand, 3 – Apply, 4 – Analyse 5 – Evaluate, 6 – Create)				