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FIRST SEMESTER M.Sc. DEGREE EXAMINATION, NOVEMBER 2023 (Regular/Improvement/Supplementary)

CHEMISTRY

FCHE1C02-ELEMENTARY INORGANIC CHEMISTRY

Time: 3 Hours

Maximum Weightage: 30

Section A: Short answer questions. Answer any *eight* questions. Each carries *one* weightage.

- 1. Write the type of π -bond involved in ClO₄⁻. Sketch it.
- 2. What is meant by hyper valence? Give an example.
- 3. Which is expected to be a better Lewis acid: BCl₃ or B(CH₃)₃? Justify your answer.
- 4. Write Lux-Flood definition of acid and base with examples.
- 5. Elaborate on the levelling effect of solvents.
- 6. What happens when 1,2-dicarboclosodecaborane [12] is heated?
- 7. Explain the merits and demarits of liquid drop model.
- 8. What are interstitial carbides? Give an example.
- 9. Illustrating an example, explain briefly thermonuclear reactions.
- 10. What are critical size and critical mass? Explain its importance in nuclear reactions.
- 11. What is Ellingham diagram? What information do we get from this diagram?
- 12. Explain Bethe's notation of nuclear reactions.

(8 x 1 = 8 weightage)

Section B: Short essay questions. Answer any *four* questions. Each carries *three* weightage.

- 13. Briefly explain Walsh diagram.
- 14. Discuss briefly the different types of hydrogen bonding and its consequences.
- 15. Give a detailed account of HSAB principle.
- 16. How borazines are synthesized? Give an account of their chemical reactivity.
- 17. Discuss the synthesis, structures and chemical properties of S₄N₄.
- 18. Explain Pourbaix diagrams and discuss their applications.
- 19. Describe the role of HF as a non-aqueous solvent.

Section C: Essay questions. Answer any two questions. Each carries five weightage.

- 20. What are silicones? Discuss their synthesis, structure and applications.
- 21. Describe the structure and bonding in diborane. How is it synthesized? Explain its reaction with ammonia.
- 22. Write notes on:

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- i) GM counter; ii) Radiation dosimetry;
- iii) Nuclear Fission; iv) Nuclear fusion.
- 23. Give a detailed account on heteropoly and isopoly acids of tungsten and molybdenum and its applications.

(2 x 5 =10 weightage)