D1ACH2102

(2 Pages)

Name..... Reg.No.....

FIRST SEMESTER M.Sc. DEGREE EXAMINATION, NOVEMBER 2021 (Regular/Improvement/Supplementary)

CHEMISTRY FCHE1C02 - ELEMENTARY INORGANIC CHEMISTRY

Time: 3 Hours

Maximum Weightage: 30

Section A: Short answer questions. Answer any *eight* questions. Each carries *one* weightage.

- 1. What are Super acids? Mention their usages.
- 2. Explain the Drago-Wayland equation.
- 3. Which is more stable; B_2H_6 or $[B_2H_6]^{2-}$? Substantiate your answer.
- 4. Classify the following compounds into *closo/nido/arachino* structures:
 (a) C₂B₁₀H₁₂ (b) C₃B₃H₅ and (c) B₄H₁₀.
- 5. How is polythiazyl prepared? Account for its metallic character.
- 6. Write action of diborane on: (a) CO and (b) PH₃.
- 7. What is radiation dosimetry?
- 8. What are Scintillation counters? Give any two examples.
- 9. How silicones are prepared? Account for their water repellent nature.
- 10. What is Frost diagram? Explain the importance.
- 11. What is meant by hyper valence? Give example.
- 12. Explain the type of π bond involved in ClO₄⁻

(8×1 = 8 weightage)

Section B: Short essay questions. Answer any *four* questions. Each carries *three* weightage.

- 13. Write notes on Isopoly and heteropoly anions.
- 14. Describe various applications of HSAB concept.
- 15. Write briefly on metallocarboranes.
- 16. Derive *styx* code for B_4H_{10} and draw its structure.
- 17. Discuss the importance of neutron activation analysis.

- 18. What are Pourbaix diagrams? Discuss applications.
- 19. Write salient features of liquid drop model.

 $(4 \times 3 = 12 \text{ weightage})$

Section C: Essay questions. Answer any two questions. Each carries five weightage.

- 20. Discuss the magnetic and spectral properties of lanthanides and actinides.
- 21. What are phosphazines? Explain preparation, properties and structure of (PNCl₂)₃. Mention their uses.
- 22. Discuss the Bent rule and energies involved in hybridisation.
- 23. Discuss the behaviour of liquid SO₂ and liquid NH₃ as solvents.

 $(2 \times 5 = 10 \text{ weightage})$